

Dynamics of chemical reaction networks and positivity of polynomials

Angélica Torres

September 15, 2020

GOAL

Give an insight on how studying the positivity of a polynomial $p(x_1, \dots, x_n)$ in $\mathbb{R}_{>0}^n$, is used to detect different dynamic behaviors in a Chemical Reaction Network.

Chemical Reaction Networks

Dynamical system

Main questions

Stability of steady states

Hurwitz criterion

A second example

Some recent results

Table of Contents

Chemical Reaction Networks

Dynamical system

Main questions

Stability of steady states

Hurwitz criterion

A second example

Some recent results

What is a Chemical Reaction Network?



What is a Chemical Reaction Network?



A chemical reaction network \mathcal{G} is a directed graph.

- Nodes: Integer linear combinations of a finite set $\mathcal{S} = \{X_1, \dots, X_n\}$

What is a Chemical Reaction Network?



A chemical reaction network \mathcal{G} is a directed graph.

- Nodes: Integer linear combinations of a finite set $\mathcal{S} = \{X_1, \dots, X_n\}$
- Edges: Interactions among species. No loops.

What is a Chemical Reaction Network?



A chemical reaction network \mathcal{G} is a directed graph.

- Nodes: Integer linear combinations of a finite set $\mathcal{S} = \{X_1, \dots, X_n\}$
- Edges: Interactions among species. No loops.

We study the change of concentration of the species, through time.

What is a Chemical Reaction Network?



What is a Chemical Reaction Network?



In this example $n = 4$ and $\mathcal{S} = \{X_1, X_2, X_3, X_4\}$.

What is a Chemical Reaction Network?



What happens when each reaction occurs?

What is a Chemical Reaction Network?



What happens when each reaction occurs?

$$\begin{pmatrix} -1 \\ 1 \\ 0 \\ 0 \end{pmatrix}$$

What is a Chemical Reaction Network?



What happens when each reaction occurs?

$$\begin{pmatrix} -1 \\ 1 \\ 0 \\ 0 \end{pmatrix}$$

$$\begin{pmatrix} 1 \\ -1 \\ -1 \\ 1 \end{pmatrix}$$

$$\begin{pmatrix} 0 \\ 0 \\ 1 \\ -1 \end{pmatrix}$$

What is a Chemical Reaction Network?



What happens when each reaction occurs?

$$N_1 = \begin{pmatrix} -1 \\ 1 \\ 0 \\ 0 \end{pmatrix}$$

$$N_2 = \begin{pmatrix} 1 \\ -1 \\ -1 \\ 1 \end{pmatrix}$$

$$N_3 = \begin{pmatrix} 0 \\ 0 \\ 1 \\ -1 \end{pmatrix}$$

The subspace $S = \text{Span}\{N_1, N_2, N_3\}$ is called the stoichiometric subspace.

System of ODEs



Under mass action kinetics, we assume that the rate of occurrence of each reaction is proportional to the product of the species in the origin of the reaction.

System of ODEs



Under mass action kinetics, we assume that the rate of occurrence of each reaction is proportional to the product of the species in the origin of the reaction.

$$v_1 = k_1 X_1 \quad v_2 = k_2 X_2 X_3 \quad v_3 = k_3 X_4.$$

System of ODEs



Under mass action kinetics, we assume that the rate of occurrence of each reaction is proportional to the product of the species in the origin of the reaction.

$$v_1 = \kappa_1 X_1 \quad v_2 = \kappa_2 X_2 X_3 \quad v_3 = \kappa_3 X_4.$$

The dynamics of the system are given by

$$\begin{pmatrix} \dot{x}_1 \\ \dot{x}_2 \\ \dot{x}_3 \\ \dot{x}_4 \end{pmatrix} = \begin{pmatrix} N_1 & N_2 & N_3 \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \\ v_3 \end{pmatrix} = \begin{pmatrix} -\kappa_1 X_1 + \kappa_2 X_2 X_3 \\ \kappa_1 X_1 - \kappa_2 X_2 X_3 \\ -\kappa_2 X_2 X_3 + \kappa_3 X_4 \\ \kappa_2 X_2 X_3 - \kappa_3 X_4 \end{pmatrix}$$

System of ODEs

The dynamics of the system are given by

$$\begin{pmatrix} \dot{x}_1 \\ \dot{x}_2 \\ \dot{x}_3 \\ \dot{x}_4 \end{pmatrix} = \begin{pmatrix} N_1 & N_2 & N_3 \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \\ v_3 \end{pmatrix} = \begin{pmatrix} -\kappa_1 x_1 + \kappa_2 x_2 x_3 \\ \kappa_1 x_1 - \kappa_2 x_2 x_3 \\ -\kappa_2 x_2 x_3 + \kappa_3 x_4 \\ \kappa_2 x_2 x_3 - \kappa_3 x_4 \end{pmatrix}$$

The dynamics of the system are given by

$$\begin{pmatrix} \dot{x}_1 \\ \dot{x}_2 \\ \dot{x}_3 \\ \dot{x}_4 \end{pmatrix} = \begin{pmatrix} N_1 & N_2 & N_3 \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \\ v_3 \end{pmatrix} = \begin{pmatrix} -\kappa_1 x_1 + \kappa_2 x_2 x_3 \\ \kappa_1 x_1 - \kappa_2 x_2 x_3 \\ -\kappa_2 x_2 x_3 + \kappa_3 x_4 \\ \kappa_2 x_2 x_3 - \kappa_3 x_4 \end{pmatrix}$$

In general...

- N is a $n \times m$ matrix, where m is the amount of reactions. Its column span is the stoichiometric subspace.

The dynamics of the system are given by

$$\begin{pmatrix} \dot{x}_1 \\ \dot{x}_2 \\ \dot{x}_3 \\ \dot{x}_4 \end{pmatrix} = \begin{pmatrix} N_1 & N_2 & N_3 \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \\ v_3 \end{pmatrix} = \begin{pmatrix} -\kappa_1 x_1 + \kappa_2 x_2 x_3 \\ \kappa_1 x_1 - \kappa_2 x_2 x_3 \\ -\kappa_2 x_2 x_3 + \kappa_3 x_4 \\ \kappa_2 x_2 x_3 - \kappa_3 x_4 \end{pmatrix}$$

In general...

- N is a $n \times m$ matrix, where m is the amount of reactions. Its column span is the stoichiometric subspace.
- The function $f(x) = Nv$ is called the rate function.

The dynamics of the system are given by

$$\begin{pmatrix} \dot{x}_1 \\ \dot{x}_2 \\ \dot{x}_3 \\ \dot{x}_4 \end{pmatrix} = \begin{pmatrix} N_1 & N_2 & N_3 \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \\ v_3 \end{pmatrix} = \begin{pmatrix} -\kappa_1 x_1 + \kappa_2 x_2 x_3 \\ \kappa_1 x_1 - \kappa_2 x_2 x_3 \\ -\kappa_2 x_2 x_3 + \kappa_3 x_4 \\ \kappa_2 x_2 x_3 - \kappa_3 x_4 \end{pmatrix}$$

In general...

- N is a $n \times m$ matrix, where m is the amount of reactions. Its column span is the stoichiometric subspace.
- The function $f(x) = Nv$ is called the rate function.
- The reaction rate constants $\kappa_1, \dots, \kappa_m$ are positive.

Steady states

The steady states are the non-negative points where the vector of derivatives of the concentrations is zero.

$$\dot{x} = 0 = f(x).$$

The positive steady states are the solutions to $0 = f(x)$ that are in the positive orthant.

Steady states

The steady states are the non-negative points where the vector of derivatives of the concentrations is zero.

$$\dot{x} = 0 = f(x).$$

The positive steady states are the solutions to $0 = f(x)$ that are in the positive orthant.

In our example

$$0 = -\kappa_1 x_1 + \kappa_2 x_2 x_3$$

$$0 = \kappa_1 x_1 - \kappa_2 x_2 x_3,$$

$$0 = -\kappa_2 x_2 x_3 + \kappa_3 x_4$$

$$0 = \kappa_2 x_2 x_3 - \kappa_3 x_4$$

Steady states

The steady states are the non-negative points where the vector of derivatives of the concentrations is zero.

$$\dot{x} = 0 = f(x).$$

The positive steady states are the solutions to $0 = f(x)$ that are in the positive orthant.

In our example

$$\phi(x_2, x_4) = \left(\frac{\kappa_3 x_4}{\kappa_1}, x_2, \frac{\kappa_3 x_4}{\kappa_2 x_2}, x_4 \right). \quad (1)$$

Given an initial solution x_0 for the system of ODEs, the trajectories containing x_0 , remain in $x_0 + S$ (*Compatibility class*). Therefore, we study the dynamics of the network within $x_0 + S$.

Given an initial solution x_0 for the system of ODEs, the trajectories containing x_0 , remain in $x_0 + S$ (*Compatibility class*). Therefore, we study the dynamics of the network within $x_0 + S$.

Conservation laws

Equations given by

$$Wx = Wx_0$$

where the rows of W form a basis of S^\perp .

Given an initial solution x_0 for the system of ODEs, the trajectories containing x_0 , remain in $x_0 + S$ (*Compatibility class*). Therefore, we study the dynamics of the network within $x_0 + S$.

Conservation laws

Equations given by

$$Wx = Wx_0$$

where the rows of W form a basis of S^\perp .

The vector Wx_0 is denoted by T and is called the *vector of total amounts*.

Chemical Reaction Networks

In our example

$$\dot{x}_1 = -\kappa_1 x_1 + \kappa_2 x_2 x_3$$

$$\dot{x}_2 = \kappa_1 x_1 - \kappa_2 x_2 x_3,$$

$$\dot{x}_3 = -\kappa_2 x_2 x_3 + \kappa_3 x_4$$

$$\dot{x}_4 = \kappa_2 x_2 x_3 - \kappa_3 x_4$$

Conservation laws

$$\begin{pmatrix} 1 & 1 & 0 & 0 \\ 0 & 0 & 1 & 1 \end{pmatrix} \begin{pmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \end{pmatrix} = \begin{pmatrix} T_1 \\ T_2 \\ T_3 \\ T_4 \end{pmatrix}.$$

Chemical Reaction Networks

In our example

$$\dot{x}_1 = -\kappa_1 x_1 + \kappa_2 x_2 x_3$$

$$\dot{x}_2 = \kappa_1 x_1 - \kappa_2 x_2 x_3,$$

$$\dot{x}_3 = -\kappa_2 x_2 x_3 + \kappa_3 x_4$$

$$\dot{x}_4 = \kappa_2 x_2 x_3 - \kappa_3 x_4$$

Conservation laws

$$x_1 + x_2 = T_1 \quad x_3 + x_4 = T_2$$

Some interesting questions about CRNs

- Given a set of parameters, is there a positive steady state?

Some interesting questions about CRNs

- Given a set of parameters, is there a positive steady state?
- Is there a set of parameters such that there is more than one steady state in a compatibility class?

Some interesting questions about CRNs

- Given a set of parameters, is there a positive steady state?
- Is there a set of parameters such that there is more than one steady state in a compatibility class?
- Given a steady state, is it asymptotically stable?

Some interesting questions about CRNs

- Given a set of parameters, is there a positive steady state?
- Is there a set of parameters such that there is more than one steady state in a compatibility class?
- Given a steady state, is it asymptotically stable?
- Is there a parameter region where the network has more than one stable steady state?

Some interesting questions about CRNs

- Given a set of parameters, is there a positive steady state?
- Is there a set of parameters such that there is more than one steady state in a compatibility class?
- Given a steady state, is it asymptotically stable?
- Is there a parameter region where the network has more than one stable steady state?
- Are all the positive steady states contained in a hyperplane parallel to $x_i = 0$ for some i ?

Some interesting questions about CRNs

- Given a set of parameters, is there a positive steady state?
- Is there a set of parameters such that there is more than one steady state in a compatibility class?
- Given a steady state, is it asymptotically stable?
- Is there a parameter region where the network has more than one stable steady state?
- Are all the positive steady states contained in a hyperplane parallel to $x_i = 0$ for some i ?

Table of Contents

Chemical Reaction Networks

Dynamical system

Main questions

Stability of steady states

Hurwitz criterion

A second example

Some recent results

STABILITY

Consider a system of differential equations $\frac{dx}{dt} = f(x)$, with $f \in \mathcal{C}^1$,
and a steady state x^* .

The steady state x^* is *asymptotically stable* if all the eigenvalues of $J_f(x^*)$ have negative real part. If one of the eigenvalues of $J_f(x^*)$ has positive real part, then x^* is *unstable*.

Stability: Hurwitz criterion.

Let $p(x) = a_n x^n + a_{n-1} x^{n-1} + \dots + a_1 x + a_0$ be a polynomial with $a_i \in \mathbb{R}$, $a_n > 0$ and $a_0 \neq 0$. The Hurwitz matrix associated to p is

$$H = \begin{pmatrix} a_{n-1} & a_n & 0 & 0 & \cdots & 0 \\ a_{n-3} & a_{n-2} & a_{n-1} & a_n & \cdots & 0 \\ \vdots & \vdots & \vdots & \vdots & \vdots & \vdots \\ 0 & 0 & 0 & a_{6-n} & \cdots & a_2 \\ 0 & 0 & 0 & 0 & \cdots & a_0 \end{pmatrix}$$

The i -th Hurwitz determinant, is $H_i = \det(H_{I,I})$, with $I = \{1, \dots, i\}$.

Stability: Hurwitz criterion.

Let $p(x) = a_n x^n + a_{n-1} x^{n-1} + \dots + a_1 x + a_0$ be a polynomial with $a_i \in \mathbb{R}$, $a_n > 0$ and $a_0 \neq 0$. The Hurwitz matrix associated to p is

$$H = \begin{pmatrix} a_{n-1} & a_n & 0 & 0 & \cdots & 0 \\ a_{n-3} & a_{n-2} & a_{n-1} & a_n & \cdots & 0 \\ \vdots & \vdots & \vdots & \vdots & \vdots & \vdots \\ 0 & 0 & 0 & a_{6-n} & \cdots & a_2 \\ 0 & 0 & 0 & 0 & \cdots & a_0 \end{pmatrix}$$

The i -th Hurwitz determinant, is $H_i = \det(H_{I,I})$, with $I = \{1, \dots, i\}$.

All the roots of the polynomial p have negative real part if, and only if, $H_i > 0$ for $i = 1, \dots, n$. If $H_i < 0$ for some i , then p has a root with positive real part.

Stability: Hurwitz criterion.

Let $p(x) = a_n x^n + a_{n-1} x^{n-1} + \dots + a_1 x + a_0$ be a polynomial with $a_i \in \mathbb{R}$, $a_n > 0$ and $a_0 \neq 0$. The Hurwitz matrix associated to p is

$$H = \begin{pmatrix} a_{n-1} & a_n & 0 & 0 & \cdots & 0 \\ a_{n-3} & a_{n-2} & a_{n-1} & a_n & \cdots & 0 \\ \vdots & \vdots & \vdots & \vdots & \vdots & \vdots \\ 0 & 0 & 0 & a_{6-n} & \cdots & a_2 \\ 0 & 0 & 0 & 0 & \cdots & a_0 \end{pmatrix}$$

The i -th Hurwitz determinant, is $H_i = \det(H_{I,I})$, with $I = \{1, \dots, i\}$.

All the roots of the polynomial p have negative real part if, and only if, $H_i > 0$ for $i = 1, \dots, n$. If $H_i < 0$ for some i , then p has a root with positive real part.

Example

In our example

$$X_1 \xrightarrow{\kappa_1} X_2 \quad X_2 + X_3 \xrightarrow{\kappa_2} X_1 + X_4 \quad X_4 \xrightarrow{\kappa_3} X_3.$$

We compute J_f and evaluate at $\phi(x_2, x_4)$.

$$J_f(\phi(x_2, x_4)) = \begin{pmatrix} -\kappa_1 & \frac{\kappa_3 x_4}{x_2} & \kappa_2 x_2 & 0 \\ \kappa_1 & -\frac{\kappa_3 x_4}{x_2} & -\kappa_2 x_2 & 0 \\ 0 & -\frac{\kappa_3 x_4}{x_2} & -\kappa_2 x_2 & \kappa_3 \\ 0 & \frac{\kappa_3 x_4}{x_2} & \kappa_2 x_2 & -\kappa_3 \end{pmatrix}$$

Example

The characteristic polynomial of $J_f(x^*)$, $p_{J_f}(\lambda)$, and factor λ^d , where d is the amount of conservation laws.

$$\begin{aligned} p_{J_f}(\lambda) &= \lambda^4 + \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2} \lambda^3 + \frac{\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4}{x_2} \lambda^2 \\ &= \lambda^2 \left(\lambda^2 + \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2} \lambda + \frac{\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4}{x_2} \right) \end{aligned}$$

Example

The polynomial we have to study is.

$$q_f(\lambda) = \lambda^2 + \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2} \lambda + \frac{\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4}{x_2}$$

Example

The polynomial we have to study is.

$$q_f(\lambda) = \lambda^2 + \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2} \lambda + \frac{\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4}{x_2}$$

The Hurwitz determinants are

$$H_1 = \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2}$$

$$H_2 = \frac{(\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4)(\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4)}{x_2^2}$$

Example

The polynomial we have to study is.

$$q_f(\lambda) = \lambda^2 + \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2} \lambda + \frac{\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4}{x_2}$$

The Hurwitz determinants are

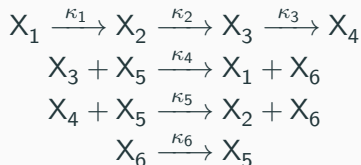
$$H_1 = \frac{\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4}{x_2}$$

$$H_2 = \frac{(\kappa_2 x_2^2 + \kappa_1 x_2 + \kappa_3 x_2 + \kappa_3 x_4)(\kappa_1 \kappa_2 x_2^2 + \kappa_1 \kappa_3 x_2 + \kappa_3^2 x_4)}{x_2^2}$$

For all choice of reaction rate constants and total amounts, the steady state is **asymptotically stable**.

A second example

Consider the following network



In this case the characteristic polynomial of $J_f(\varphi)$, has degree 4 which means that we have to analyse 4 Hurwitz determinants.

A second example

H_1, H_2 and H_3 are polynomials with positive coefficients.

$$H_4 = \frac{1}{\kappa_3} \left((\kappa_1 + \kappa_2) \kappa_3 \kappa_4 \kappa_5 \kappa_6 x_5^2 + \kappa_2 \kappa_4 \kappa_5^2 (\kappa_1 - \kappa_3) x_4 x_5^2 \right. \\ \left. + 2 \kappa_2 \kappa_3 \kappa_4 \kappa_5 x_4 x_5 + \kappa_1 (\kappa_2 + \kappa_3) \kappa_3 \kappa_5 \kappa_6 x_5 + \kappa_1 \kappa_2 \kappa_3^2 \kappa_5 x_4 + \right. \\ \left. \kappa_1 \kappa_2 \kappa_3^2 \kappa_6 \right) H_3$$

Stability depends on the sign of the first factor.

A second example

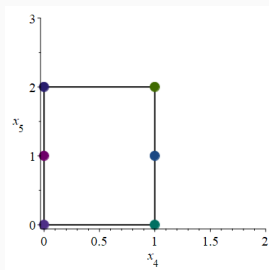
The Newton polytope of

$$p(x_4, x_5) = \frac{1}{\kappa_3} \left((\kappa_1 + \kappa_2) \kappa_3 \kappa_4 \kappa_5 \kappa_6 x_5^2 + \kappa_2 \kappa_4 \kappa_5^2 (\kappa_1 - \kappa_3) x_4 x_5^2 \right. \\ \left. + 2 \kappa_2 \kappa_3 \kappa_4 \kappa_5 x_4 x_5 + \kappa_1 (\kappa_2 + \kappa_3) \kappa_3 \kappa_5 \kappa_6 x_5 + \kappa_1 \kappa_2 \kappa_3^2 \kappa_5 x_4 + \right. \\ \left. \kappa_1 \kappa_2 \kappa_3^2 \kappa_6 \right)$$

A second example

The Newton polytope of

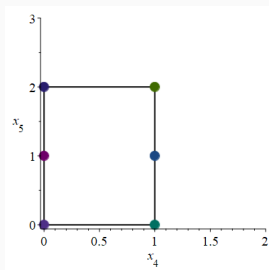
$$p(x_4, x_5) = \frac{1}{\kappa_3} \left((\kappa_1 + \kappa_2) \kappa_3 \kappa_4 \kappa_5 \kappa_6 x_5^2 + \kappa_2 \kappa_4 \kappa_5^2 (\kappa_1 - \kappa_3) x_4 x_5^2 \right. \\ \left. + 2\kappa_2 \kappa_3 \kappa_4 \kappa_5 x_4 x_5 + \kappa_1 (\kappa_2 + \kappa_3) \kappa_3 \kappa_5 \kappa_6 x_5 + \kappa_1 \kappa_2 \kappa_3^2 \kappa_5 x_4 + \kappa_1 \kappa_2 \kappa_3^2 \kappa_6 \right)$$



A second example

The Newton polytope of

$$p(x_4, x_5) = \frac{1}{\kappa_3} \left((\kappa_1 + \kappa_2) \kappa_3 \kappa_4 \kappa_5 \kappa_6 x_5^2 + \kappa_2 \kappa_4 \kappa_5^2 (\kappa_1 - \kappa_3) x_4 x_5^2 \right. \\ \left. + 2 \kappa_2 \kappa_3 \kappa_4 \kappa_5 x_4 x_5 + \kappa_1 (\kappa_2 + \kappa_3) \kappa_3 \kappa_5 \kappa_6 x_5 + \kappa_1 \kappa_2 \kappa_3^2 \kappa_5 x_4 + \kappa_1 \kappa_2 \kappa_3^2 \kappa_6 \right)$$



If $\kappa_3 > \kappa_1$ it is possible to find one unstable steady state. Otherwise, all the steady states are asymptotically stable.

What do we need for bigger networks?

When using algebraic parameterizations, computing the Hurwitz determinants and studying their sign is computationally expensive.

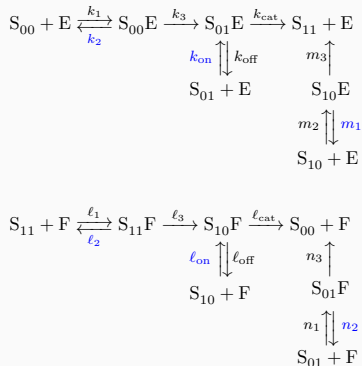


Figure 1: Model of ERK regulation

Table of Contents

Chemical Reaction Networks

- Dynamical system

- Main questions

Stability of steady states

- Hurwitz criterion

- A second example

Some recent results

Model or ERK regulation

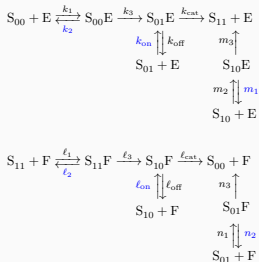


Figure 2: Model of ERK regulation

N. Obatake, A. Shiu, X. Tang, .

- There is a set of parameters where the model of ERK regulation is bistable.
- Bistability arises only if $\kappa_{on} > 0$ or $\ell_{on} > 0$.

Bistability in a CRN

	Network	Reduced network
(a)	$S_0 + E \rightleftharpoons S_0E \rightarrow S_1 + E$	$S_0 + E \rightarrow S_0E \rightarrow S_1 + E$
	$S_1 + E \rightleftharpoons S_1E \rightarrow S_2 + E$	$S_1 + E \rightarrow S_1E \rightarrow S_2 + E$
	$S_1 + F_1 \rightleftharpoons S_1F_1 \rightarrow S_0 + F_1$	$S_1 + F_1 \rightarrow S_0 + F_1$
	$S_2 + F_2 \rightleftharpoons S_2F_2 \rightarrow S_1 + F_2$	$S_2 + F_2 \rightarrow S_1 + F_2$

E. Feliu, .

Procedure to detect bistability by studying model reductions of a network.

Bistability in a CRN

	Network	Reduced network
(a)	$S_0 + E \rightleftharpoons S_0E \rightarrow S_1 + E$	$S_0 + E \rightarrow S_0E \rightarrow S_1 + E$
	$S_1 + E \rightleftharpoons S_1E \rightarrow S_2 + E$	$S_1 + E \rightarrow S_1E \rightarrow S_2 + E$
	$S_1 + F_1 \rightleftharpoons S_1F_1 \rightarrow S_0 + F_1$	$S_1 + F_1 \rightarrow S_0 + F_1$
	$S_2 + F_2 \rightleftharpoons S_2F_2 \rightarrow S_1 + F_2$	$S_2 + F_2 \rightarrow S_1 + F_2$

E. Feliu, .

Procedure to detect bistability by studying model reductions of a network.

THANK YOU!