Protein Physics 2016

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Secondary structure stability, beta-sheet formation & stability

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Recap of statistics

- Energy Entropy
- Entropy microstates volume & order
- Probability of being in a state i: $w_i(T) = \frac{\exp(-\epsilon_i/k_B T)}{Z(T)}$
- Partition function:

$$Z(T) = \sum_{i} \exp\left(-\epsilon_i / k_B T\right)$$

Alpha helix formation

- Hydrogen bonds: i to i+4
 - 0-4, 1-5, 2-6
- First hydrogen bond "locks" residues 1,2,3 in place



- Second stabilizes 2,3,4 (etc.)
- N residues stabilized by N-2 hydrogen bonds!



Alpha helix free energy

• Free energy of helix vs. "coil" states:



How does a helix form?

- Landau: Phases cannot co-exist in 3D
- First order phase transitions means either state can be stable, but not the mixture
- Think ice/water either freezing or melting $n \propto V \propto r^3$

 $A \propto r^2 \propto n^{2/3}$ Surface tension costly!

- But a helix-coil transition in a chain is 1D!
- Interface helix/coil does not depend on N!

Helix stability

- Temperature dependence
- Elongation term dominant for large N
- Why? Since it is raised to the power of N!



ighly cooperative, but NOT a formal phase transition! (width does not go to zero)

Formation...

T: I-residue Rate of formation at position 1: $t_{\text{INIT0}} = \tau \exp \left(f_{\text{INIT}} / kT \right) = \tau / \sigma$ Rate of formation anywhere $(n0 \approx 1/\sqrt{\sigma})$: $t_{\rm INIT} = \tau / \sqrt{\sigma}$ • Propagation to all residues: $tn_0 = \tau / \sqrt{\sigma}$ • Total time is ~2t_{INIT}, halftime thus ~t_{INIT}. • Half time spent on initiation, half elongation!

Helix summary

- Very fast formation
- Both initiation & elongation matters
- Quantitative values derived from CD-spectra
- Low free energy barriers, ~1kcal/mol
- Characteristic lengths 20-30 residues

Beta sheet formation

- Experimentally: Can take hours to weeks!
- But sometimes just a millisecond. Why?
- Is it initiation- or elongation-limited?
- Beta sheet formation appears to be a typical first-order phase transition!

Beta sheet formation



Sheets vs. Helices

- Beta sheets are two-dimensional
- Interface area grows with # residues
 - Phases cannot coexist and there will be a first-order phase transition
- Structure interface:
 - Sheet edges & bends/loops

Beta sheet energies

- f_β: Free energy of residue inside a single beta hairpin, relative to the random coil
- Δf_{β} : Extra edge free energy
 - Total free energy at edge is $f_{\beta+} \Delta f_{\beta}$
- U: Free energy of bend/coil per residue
- Since sheets can form we must have $\Delta f_{\beta} > 0 \& U > 0!$ Why?

Two Scenarios:

- f_{β+} Δf_β<0: A single long beta hairpin will be more stable than coil. Only a single turn required for formation
- f_{β+} Δf_β>0: Hairpins are only formed because of association with other residues into a beta sheet. Activation barrier is the formation of a sheet "nucleus"

Minimum strand length

- Consider the case when single hairpins are not stable.
- $\Delta F = U + 2N(f_{\beta} + \Delta f_{\beta})$ All residues face an edge • Association with a new strand maintains edge, and gives us N new internal resides: $\Delta F' = -Nf_{\beta}$
- Formation of next turn: $\Delta F^{\prime\prime} = U$
- Minimum strand length: $N_{\min} = U/(-f_{\beta})$

Beta transition state

• Find the highest-free-energy intermediate: Single hairpin with a following turn $F^{\#} = U + 2N_{\min}(f_{\beta} + \Delta f_{\beta}) + U = 2(U\Delta f_{\beta})/(-f_{\beta})$





The book goes into some detail to prove that this is the lowest possible transition state energy! Why is that important?

Beta formation rates

- Initiation at a given point: $t_{\text{INIT0}} \approx \tau_{\beta} \exp\left(+F^{\#}/kT\right)$
- Initiation somewhere: $t_{\rm INIT0}/N$
- Initiation is entirely time-limiting
- Total formation time: $t \approx \tau_{\beta} \exp(F^{\#}/kT)/N$
- And remember that we had: $F^{\#} = U + 2N_{\min}(f_{\beta} + \Delta f_{\beta}) + U = 2(U\Delta f_{\beta})/(-f_{\beta})$

Beta formation rates

- Rate depends on β -structure stability: $t_{\beta} \approx \exp \left[A/(-f_{\beta}) \right]$
- Exponential dependence on residue beta stability explains wide range of formation times observed in experiments!



Beta sheet summary

- Unstable sheets are extremely slow to form (hours to weeks)
- Stable sheets can form in milliseconds
- Significant free energy barrier
- Beta sheet folding is a first-order phase transition

Helix-sheet comparison

- The alpha helix "avoids" the phase transition - the boundary area does not increase with helix size
- Leads to much lower barriers, which can be overcome in microseconds
- The high free energy barrier of sheets is likely one of the explanations to prion/ amyloid protein misfolding diseases

Misfolding - Prions



Misfolded form (right) is protease resistant

The *in vivo* state is only the second best here - but the free energy barrier to the lowest is very high!



LAG PHASE

GROWTH PHASE



What is the Coil?

- Less well-defined state than native
- It is NOT a stretched out linear chain!



What is the average chain $\mathbf{h} = \sum_{i=1}^{N} \mathbf{r}_{i}$ end-to-end distance?



Average coil length

2

$$egin{array}{rl} \mathbf{h}^2
ight
angle &= \left\langle \left(\sum_{i=1}^N \mathbf{r}_i
ight)^2
ight. \ &= \left\langle \sum_{i=1}^N \mathbf{r}_i^2 + \sum_{i=1}^N \sum_{j
eq i}^N \mathbf{r}_i \mathbf{r}_j
ight
angle \ &= \sum_{i=1}^N \left\langle \mathbf{r}_i^2
ight
angle + \sum_{i=1}^N \sum_{j
eq i}^N \left\langle \mathbf{r}_i \mathbf{r}_j
ight
angle = N \end{array}$$

Average length increases as \sqrt{N} Average volume increases as $N^{3/2}$

Average coil length

- Some problems:
 - Segments cannot have any orientation
 - Angle potentials
- Generalized expression: $\langle \mathbf{h}^2 \rangle = Nr^2 = Lr$
- Rotational model:

 $r = l\left(1 + \left<\cos\alpha\right>\right) / \left(1 - \left<\cos\alpha\right>\right)$



Chain contour

Excluded volume

- Real chains cannot cross themselves!
- Segment i can never overlap with j, even if they are very far apart
- Excluded volume effects!

 $\left|\sqrt{\left<\mathbf{h}^2\right>}\approx N^{0.588}r\right|$



Summary

- Alpha helix & beta sheets form in very different ways, that give them different properties!
- All determined by free energy barriers!
- There are natural sizes of helices/sheets
- Folding rates can be predicted with very simple qualitative arguments
- You should understand both how & *why* the are different (i.e. be able to explain)!