## Environmental Science, Problems Chapter 4

## 4.1

Assuming a $\mathrm{CO}_{2}$ volume fraction of 350 ppmv , calculate the weight fraction of $\mathrm{CO}_{2}$. The average density of air is $1.29 \mathrm{~kg} / \mathrm{m}^{3}$ and the density of $\mathrm{CO}_{2}$ is $1.98 \mathrm{~kg} / \mathrm{m}^{3}$ (STP).

Answer: 537 ppm (weight)

## 4.2

The seasonal variations in atmospheric $\mathrm{CO}_{2}$ amount to about 4 ppmv. Estimate the total volume needed to store this amount of $\mathrm{CO}_{2}$ in liquid phase (density $770 \mathrm{~kg} / \mathrm{m}^{3}$ ).

Answer: $1.1 * 10^{4} \mathrm{~km}^{3}$

