



# KD1230 Organic Chemistry, Basic Concepts and Practice 6.0 credits

Organisk kemi, grundläggande koncept och praktik

This is a translation of the Swedish, legally binding, course syllabus.

## Establishment

Course syllabus for KD1230 valid from Spring 2019

## Grading scale

A, B, C, D, E, FX, F

## Education cycle

First cycle

## Main field of study

Technology

## Specific prerequisites

Basic eligibility and special eligibility are required in Mathematics course E, Physics course B, Chemistry course A. In each and every of these subjects, the lowest accepted grade is passed or 3.

Recommended prerequisites: KE1140 Engineering Chemistry.

# Language of instruction

The language of instruction is specified in the course offering information in the course catalogue.

## Intended learning outcomes

After completing the course the student should be able to:

- apply the organic chemistry language, e.g. describe organic structures graphically, naming organic compounds according to the IUPAC nomenclature, give trivial names for some common compounds and describe the three-dimensional structure of organic compounds graphically and according to CIP nomenclature.
- identify and rank nucleophiles, electrophiles, acids and bases in a chemical reaction.
- determine how acid/base equilibrium/pK /proton transfer influence the outcome of a reaction.
- use electron (arrow) pushing to describe mechanisms.
- use frontier orbital theory to categorize which orbitals that are HOMO and LUMO in organic molecules and apply these to determine the outcome of a reaction.
- motivate based on reaction conditions if a reaction proceeds through a SN1, SN2, E1 or E2 mechanism and explain the stereochemical and regiochemical outcome. Inversely, describe reaction conditions to control a reaction so that it proceeds via a SN1, SN2, E1 or E2 mechanism.
- analyze an organic compound's stereochemistry to determine its conformation and reactivity.
- explain the mechanism and the stereo- and regiochemical outcome of additions of electrophiles to alkenes.
- describe the concept of Green Chemistry and its application in organic chemistry and how this can apply for sustainable development.
- apply basic spectroscopic methods (NMR) for structural analysis of organic compounds.
- perform risk analysis and safety assessment of an organic chemical reaction process and understand the safety precautions required for laboratory work.
- turn a recipe into a complete synthesis, which include: setting up the reaction, do work up and purification of the desired compound by means of extraction, distillation and crystallization.
- structure determination of organic compounds using the most common analytical methods (melting point, NMR, IR)
- basic synthesis planning which includes; extracting relevant information from organic chemistry databases, design and execute an organic reaction process and report the result.

## Course contents

The course objective is to provide a strong introduction to organic chemistry in terms of structure and reactivity, practical synthetic work and green chemistry.

The course also provides a solid base for further specialization in organic chemistry as well as in adjacent areas, especially environmental chemistry, biochemistry, polymer technology and physical chemistry.

The course includes:

- Nomenclature
- Conformation/configuration
- Mechanisms
- Frontier Orbital Theory
- Proton Transfer/pK
- Substitution/Elimination
- Addition reactions
- Green Chemistry

## Course literature

A. Burrows, A. Parsons, G. Price; Chemistry3: Introducing Inorganic, Organic and Physical Chemistry, Oxford University Press, USA, 2009, ISBN10: 0199277893 ISBN13: 9780199277896

Säkerhetskompodium, Organisk kemi, KTH

Laborationskompodium, Organisk kemi, KTH

## Examination

- TEN1 - Written exam, 3.0 credits, grading scale: A, B, C, D, E, FX, F
- LAB1 - Laboratory Work, 3.0 credits, grading scale: P, F

Based on recommendation from KTH's coordinator for disabilities, the examiner will decide how to adapt an examination for students with documented disability.

The examiner may apply another examination format when re-examining individual students.

If the course is discontinued, students may request to be examined during the following two academic years.

LAB1 - Laboration, 3 hp, grade scale: P, F

TEN1 - Written exam, 3 hp, grade scale: A, B, C, D, E, FX, F

## Other requirements for final grade

Passed examination (TEN1) and passed laboratory work (LAB1)

## Ethical approach

- All members of a group are responsible for the group's work.
- In any assessment, every student shall honestly disclose any help received and sources used.
- In an oral assessment, every student shall be able to present and answer questions about the entire assignment and solution.