



KD1260 Chemistry of Materials

7.5 credits

Materialens kemi

This is a translation of the Swedish, legally binding, course syllabus.

If the course is discontinued, students may request to be examined during the following two academic years

Establishment

Course syllabus for KD1260 valid from Spring 2023

Decision to discontinue this course

The course will be discontinued at the end of Spring 2022 according to school head decision: C-2022-1661. The last opportunity for examination in the course is given in Spring 2024.

Grading scale

A, B, C, D, E, FX, F

Education cycle

First cycle

Main field of study

Technology

Specific prerequisites

Language of instruction

The language of instruction is specified in the course offering information in the course catalogue.

Intended learning outcomes

After graduating from the course the students should be able to:

1. Explain concepts and solve problems related to atomic structure and chemical bonds.
2. Balance chemical equations and solve stoichiometric problems
3. Explain thermodynamic concepts and solve thermodynamic problems
4. Discuss materials from a chemical perspective
5. Perform elementary chemical laboratory work with regard to the work environment, safety instructions, and safe handling of chemicals as well as compile and communicate the results in writing in a clear manner.
6. Assimilate and reflect on information from a scientific context and summarize it for a specific target group.

Course contents

The aim of this course is to provide a broad introduction to the field of chemistry in order to obtain a deeper understanding of the structure of materials and their applications. The course is a solid basis for further studies within related fields especially thermodynamics, polymer technology, and metallic materials.

The course contains

- Stoichiometry: chemical equations, oxidation and reduction, chemical reactions
- Chemical structure: atomic and molecular orbitals, chemical bonding, Lewis structures, shape of molecules
- Thermodynamics: energy and enthalpy, heat capacity, entropy, Gibbs energy

Examination

- INL1 - Assignment, 2.0 credits, grading scale: A, B, C, D, E, FX, F
- LAB1 - Laboratory Work, 2.5 credits, grading scale: P, F
- TEN1 - Written exam, 3.0 credits, grading scale: A, B, C, D, E, FX, F

Based on recommendation from KTH's coordinator for disabilities, the examiner will decide how to adapt an examination for students with documented disability.

The examiner may apply another examination format when re-examining individual students.

Ethical approach

- All members of a group are responsible for the group's work.
- In any assessment, every student shall honestly disclose any help received and sources used.
- In an oral assessment, every student shall be able to present and answer questions about the entire assignment and solution.